## NEW HYDROPHOBIC HOST MOLECULES CONTAINING MULTIPLE REDOX-ACTIVE CENTRES

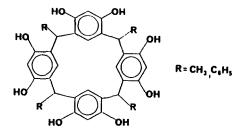
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## Abstract

The synthesis of two novel macrocyclic hydrophobic host molecules (4) and (6) containing respectively four and twelve ferrocene redox-active centres is described.

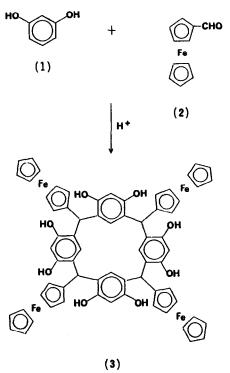
Although there is considerable current interest in the design and synthesis of receptor molecules containing a redox-active centre in close proximity to a crown ether<sup>1-3</sup> or cryptand<sup>4</sup> coordination site, few examples of the incorporation of redox centres into hydrophobic host molecules have been reported.<sup>5-7</sup> Interest in these latter molecules stems from the idea of investigating the potential catalytic interactions between the redox-active moiety and an included organic guest substrate. This paper reports the preparation of a new redox-active hydrophobic host molecule (4) that contains four ferrocenyl groups and a novel twelve ferrocenyl group containing analogue (6).

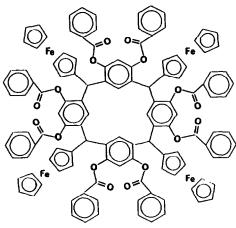
It is well known that reactions of resorcinol with aldehydes under acidic conditions can lead to macrocyclic products of type  $I.^{8,9}$  A simple adaptation of this methodology by using a metallocenecarboxaldehyde leads to



(1)

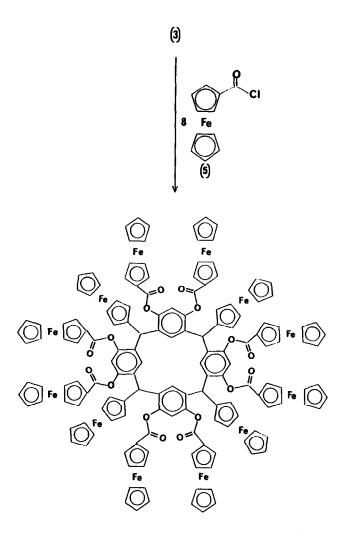
respective two step preparations of (4) and (6). The reaction of ferrocenecarboxaldehyde (1) and resorcinol (2) in the presence of hydrochloric acid and ethanol gave initially a black precipitate, insoluble in all common organic solvents tentatively assigned as the phenolic macrocycle (3). Subsequent benzoylation of a suspension of (3) in dry tetrahydrofuran followed by column chromatographic separation gave (4) as an orange crystalline solid, m.p. >250 (10% overall yield for the two steps).





(4)

An analogous procedure using ferrocenecarbonyl chloride (5) gave (6) as an orange-red crystalline solid m.p. >250°C (10% overall yield).



(6)

Elemental analyses, mass spectrometry [(4) m/z = 2057, (6) m/z 2921] and <sup>1</sup>H and <sup>13</sup>C n.m.r. spectroscopy confirmed the proposed structures.<sup>10,11</sup> It is noteworthy that the <sup>1</sup>H n.m.r. spectrum of (4) in CDCl<sub>3</sub> at ambient temperature reveals two types of ferrocene groups suggesting the solution structure of (4) possesses a C<sub>2</sub> axis of symmetry.<sup>10,11</sup> Preliminary electrochemical experiments (acetonitrile, S.C.E.) on (4) shows two, two electron reversible oxidation waves at +0.575 V and +0.665 V corresponding to the oxidation of the respective four ferrocenyl moleties.

These interesting electrochemical observations indicate the future possibility for preparing mixed-valence species which may exhibit exciting new electronic, optical, catalytic and physical properties.

## Acknowledgements

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## References

- (a) P.D. Beer, J. Chem. Soc. Chem. Commun. 1985, 1115. (b) T. Saji, Chem. Lett. 1986, 275. (c) B. Czech and A. Ratajczak, Pol. J. Chem. 1980, <u>54</u>, 767. (d) P.D. Beer and A.D. Keefe, J. Organometal. Chem. 1986, <u>306</u>, C10. (e) S. Akabori, Y. Habata, Y. Sakamoto, M. Sato and S. Ebine, Bull. Soc. Jpn. 1983, <u>56</u>, 537.
- R.E. Wolf and S.R. Cooper, J. Am. Chem. Soc. 1984, <u>106</u>, 4646. (b) L. Echegoyen, D.A. Gustowski, Soc. Chem. Commun. 1986, 220. (c) K. Maruyama, H. Sohmiya and H. Tsukube, J. Chem. Soc. Perkin Trans 1 1986, 2069. (d) F. Dietl, G. Giener and A. Merz, Synthesis 1986, 626.
- A. Kaifer, D.A. Gustowski, L. Echegoyen, V.J. Gatto, R.A. Schultz, T.P. Cleary, C.R. Morgan, D.M. Goli, A.M. Rios and C.W. Gokel, J. Am. Chem. Soc. 1985, <u>107</u>, 1958.
- (a) P.D. Beer, A.D. Keefe, C.G. Crane, A.R. Whyman, J. Organometal. Chem. 1986, <u>314</u>, C9. (b)
  P.D. Beer, C.D. Bush and T.A. Hamor, J. Organometal. Chem. 1988, <u>339</u>, 133.
- 5. A. Veno, F. Monivaki, T. Osa, F. Hamadal and K. Murai, Chem. Pharm. Bull. 1986, 34, 438.
- 6. I. Tabushi, N. Shimizu, K. Yamamura, J. Am. Chem. Soc. 1977, 99, 7100.
- 7. P.D. Beer, A.D. Keefe, J. Inclusion Phenom. 1987, 5, 499.
- 8. A.G.S. Högberg, J. Org. Chem. 1980, 45, 4498.
- 9. A.G.S. Högberg, J. Am. Chem. Soc. 1980, 102, 6046.
- The result of a single crystal X-ray analysis of (4) is in accordance with the proposed structure.
  P.D. Beer, E.L. Tite, A.M.Z. Slawin and D.J. Williams, to be reported elsewhere.
- 11. 'H NMR (CDCl<sub>3</sub>) for (4) 3.93 (24H,  $\underline{s}$ ), 4.05 (4H  $\underline{s}$ ), 4.12 (4H  $\underline{s}$ ), 4.23 (4H,  $\underline{s}$ ), 5.65 (4H, s), 6.64 (2H, s), 6.73 (2H, s), 6.75 (2H, s), 7.20-7.25 (8H, m), 7.42-7.58 (16H m), 7.82-7.99 (16H, m). P (C = 0) KBr disc = 1735 cm<sup>-1</sup>.

C<sub>124</sub>H<sub>88</sub>O<sub>16</sub>Fe<sub>4</sub> requires C, 72.4, H 4.3

found C, 72.3, H 4.5

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